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A simultaneous approach for calibrating Rate Based Models of packed distillation columns based on multiple experiments

J. Bonilla^{a,b}, F. Logist^{a,*}, B. Huyck^{a,b}, J. Degrève^a, B. De Moor^b, J. Van Impe^a

Abstract

This paper presents the calibration of a steady state Rate Based Model (RBM) for distillation based on multiple experiments. A packed column is considered, using non-equilibrium stages to represent the packing segments. For an efficient and accurate calibration, the number of model equations and parameters is first reduced via analytical manipulations and a sensitivity analysis. Second, the model is formulated such that a sparse and banded Jacobian can be exploited. Finally, novel constraints on the physical parameters are derived such that the parameter estimation yields consistent results. The model prediction capabilities are successfully validated with experimental data.

Keywords: Distillation, packed columns, parameter estimation, nonlinear programming, structure exploitation, simultaneous estimation.

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^aDepartment of Chemical Engineering CIT/BioTeC, Katholieke Universiteit Leuven, W. De Croylaan 46, B-3001 Leuven, Belgium.

^bDepartment of Electrical Engineering ESAT/SCD, Katholieke Universiteit Leuven, Kasteelpark Arenberg 10, bus 2446 B-3001 Leuven, Belgium.

^{*}Corresponding author. Tel: +32 16 322703; fax: +32 16 32299.

*Email addresses: filip.logist@cit.kuleuven.be/filip.logist@gmail.com (F. Logist), jan.vanimpe@cit.kuleuven.be (J. Van Impe)

1. Introduction

Distillation is not only the most common but also one of the most energy intensive unit operations in chemical industry. An excellent way to reduce the operating costs and energy consumption is by an appropriate (modelbased) control of, e.g., applied heat, temperatures, product compositions and flows. However, this requires models which describe accurately the plant behavior. Developing accurate first-principle models is a challenge due to the nonlinear and complex nature of the equations. Moreover, this complexity is even increased when model calibration is aimed at, which simultaneously accounts for experiments under different operating conditions. Although distillation is in industry often performed using tray columns, packed columns are in general more efficient since the vapor and liquid are continuously in contact through the packing surface, enhancing the mass and energy transfer. Packed columns can be modeled using concepts from tray columns such as stage efficiency and height equivalent of a theoretical plate (Seader and Henley, 2006). However, these concepts do not account for deviations from equilibrium properly, in contrast to Rate Based Models, which divide the packing in segments and model them as a mass and energy transfer unit (Khrishnamurthy and Taylor, 1985; Taylor and Krishna, 1993). The aim is to propose a simultaneous method for calibrating a steady-state Rate Based Model using multiple experiments and to validate it on a binary pilot-scale distillation column. A similar plant was recently studied in Barroso et al. (2009); Chambel et al. (2011). To ensure an accurate and computationally efficient result, (i) model and parameter reduction, (ii) problem structure

exploitation and (iii) additional constraints have been used.

2. Experimental context

The experimental setup involves a computer controlled packed distillation column (Figure 1). The column has an internal diameter of 7 cm containing three sections of 960 mm Sulzer CY packing. The feed stream containing methanol and isopropanol can be introduced into the column between the packed sections S_1 and S_2 or S_2 and S_3 . The feed temperature can be adjusted by an electric heater of maximum 250 W. In the reboiler located at the bottom, two electric heaters of maximum 3000 W vaporize part of the liquid. The rest is extracted as bottom stream. A total condenser at the top allows condensing the entire overhead vapor stream, which is collected in a reflux drum. Part of the condensed liquid is fed back to the column as reflux, while the remainder leaves the column as distillate. Four variables can be manipulated: the reboiler duty Q_R (W), the feed rate F (g/min), the duty of the feed heater Q_F (W) and the reflux flow rate L_N (g/min). The distillate flow D (g/min) is adjusted to maintain a constant reflux drum level. Measurements are available for the reflux flow rate L_N , the distillate flow rate D, the feed flow rate F and twelve temperatures. These temperatures are the reflux temperature $T_{s_{12}}$, the temperature at the top of the condenser $T_{s_{11}}$, the temperatures in the center and extremes of each packing section $(T_{s_2} \text{ to } T_{s_{10}})$, the temperature of the feed point T_F , and the temperature in the reboiler T_{s_1} . The concentrations in the distillate and bottom streams are measured offline by sampling.

3. Simultaneous formulation for model calibration

The steady-state model can be described by a set of nonlinear equations in the form $F(\mathbf{x},p)=0$ with \mathbf{x} the model states and p the model parameters to be calibrated. Typically, the parameter values have to be optimised such that model predictions are as close as possible to the measured outputs \mathbf{y} . In a sequential approach only the parameters are degrees of freedom in the optimization problem while the model equations are solved each time as an inner simulation problem. Alternatively, in a simultaneous approach, both the parameter and state variables are degrees of freedom in the optimization problem, while the model equations are introduced as additional equality constraints. This is a more efficient approach that preserves sparsity in the optimization, at the cost of increasing the number of optimization variables. Hence, the optimization problem is cast as:

$$\min_{\mathbf{x},p} \|\bar{\mathbf{y}} - \mathbf{C}\mathbf{x}\|_{Q_{\mathbf{x}}}^{2} \text{ subject to } \begin{cases} F(\mathbf{x},p) = 0 \\ \mathbf{x}_{\min} \leq \mathbf{x} \leq \mathbf{x}_{\max} \\ p_{\min} \leq p \leq p_{\max} \end{cases} \tag{1}$$

where the vector $\bar{\mathbf{y}}$ represents the measurement data, C is a positive semidefinite diagonal matrix with zero entries in the diagonal corresponding to the states that are not measured and Q_x is a weight matrix. Note that this formulation accounts only for one experiment. In order to use multiple experiments, the optimization vector, the residual vector and the constraints are augmented such that:

$$\tilde{\mathbf{y}}^{T} = [\bar{\mathbf{y}}_{1}^{T}, \dots, \bar{\mathbf{y}}_{M}^{T}], \quad \mathbf{w}^{T} = [\mathbf{x}_{1}^{T}, \dots, \mathbf{x}_{M}^{T}, p],$$

$$\mathbf{F}(\mathbf{w})^{T} = [F(\mathbf{x}_{1}, p)^{T}, \dots, F(\mathbf{x}_{M}, p)^{T}]$$
(2)

and the problem is formulated as:

$$\min_{\mathbf{w}} \|\tilde{\mathbf{y}} - \tilde{\mathbf{C}}\mathbf{w}\|_{Q_{\mathbf{w}}}^{2} \text{ subject to } \begin{cases} F(\mathbf{w}) = 0 \\ \mathbf{w}_{\min} \leq \mathbf{w} \leq \mathbf{w}_{\max} \end{cases}$$
(3)

with appropriate matrices $Q_{\rm w}$, $\tilde{\rm C}$, and appropriate bounds ${\rm w_{min}}$, ${\rm w_{max}}$. As the size of the optimization problems grows with the number of experiments, model reduction and structure exploitation are of major importance.

4. Numerical approach

This section details the steps taken to accurately and efficiently solve the calibration problem.

4.1. Model reduction

Starting from a general Rate Based Model (Khrishnamurthy and Taylor, 1985), a reduced order model is derived based on the following assumptions.

- A binary mixture is considered.
- Bulk phases are perfectly mixed.
- Vapor-liquid equilibrium is only valid at the vapor-liquid interface.
- The reboiler and condenser are in thermodynamical equilibrium.
- The liquid volumes of reboiler and reflux drum, \bar{v}_R and \bar{v}_D , are perfectly controlled.
- Each stage is in mechanical equilibrium.

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• Equilibrium holds in the condenser. Subcooling in the condenser is

attributed to the heat loss in the reflux drum.

The reduction is performed by inclusion of these assumptions as well as differ-

entiation of enthalpy correlations and algebraic manipulation of energy and

mass balance equations. As a result the number of variables and equations

in the Reduced Order Rate Based Model (RORBM) is now 14N - 12 instead

of 20N - 17 in the Full Rate Based Model (FRBM), with N the number of

stages (Bonilla, 2011; Bonilla et al., 2012). This means a reduction of 6N-5

equations, or approximately 30%.

4.2. Parameter reduction

To reduce the number of parameters to be estimated, a parameter sensi-

tivity analysis is performed, i.e., an analysis of the effect of the parameters on

the model states. This analysis indicates which parameters can be estimated

from the steady state measurements.

4.3. Structure exploitation

The steady state constraints give rise to a root finding problem where a

nonlinear system of equations of the form F(x, p) = 0 has to be solved. Al-

though its structure is obtained directly from the formulation of balances and

equilibrium relations separately, the variables and equations can be grouped

per stage in order to exploit sparsity. This leads to a sparse and reduced

band pattern in the model Jacobian (see Bonilla et al. (2012) for the actual

re-arranged banded structure). This banded structure is known in distilla-

tion models and equation-tearing methods have been proposed to solve the

nonlinear system of equations (Seader and Henley, 2006; Taylor and Krishna,

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1993). However, a Newton based approach yields a more general procedure

and provides more flexibility in the problem specification (Seader and Hen-

ley, 2006). Hence, the sparsity pattern is exploited (Dongarra et al., 1988)

at each Newton iteration of the preconditioned conjugate gradient algorithm

used. Proper scaling of the equations accelerates the convergence (Nocedal

and Wright, 2006).

4.4. Physically inspired constraints

Although the parameters have a physical meaning, some combinations of

their positive values have been observed to yield state profiles that are phys-

ically impossible (Bonilla, 2011). This issue can be solved by adding proper

constraints. As the current RBM does not incorporate any constraint by itself

that restrict pairs (y^V, T^V) or $(x^L, T^L)^1$ to superheated vapor or subcooled

liquid regions, respectively. This translates into additional inequality con-

straints, which require the computation of the dew and bubble point curves

and, hence, pose an embedded root finding problem. However, as a simulta-

neous approach is exploited the relations for dew and bubble points can be

solved as a part of the model equations.

5. Results and discussion

This section presents and discusses the obtained results.

 y^{V} represents the composition in the bulk vapor phase, x^{L} the composition in the

bulk liquid phase, T^V the temperature of the bulk vapor phase and T^L the temperature

of the bulk liquid phase.

5.1. Computational speedup

To illustrate the necessity of the model reduction and sparsity exploitation, a column with N equal to 20 is simulated as a test case. The computational statistics can be found in Table 2. Clearly, when going from the full to the reduced order rate based model, the number of function evaluations and computation time is roughly halved. However, when exploiting sparsity, it is seen that a 90% reduction is possible. Hence, the combined effect yields a 20 times speedup. Moreover, this effect will only grow when the size is increases, e.g., by simultaneously considering multiple experiments.

5.2. Experimental data

Five experiments are used for identification and two for validation. The values for the manipulated variables are presented in Table 3 while the measured steady state profiles for eleven temperature sensors are illustrated in Figure 7. Note that the measurement coming from the sensor in the condenser $T_{s_{11}}$ corresponds to a temperature below the boiling point of pure methanol at atmospheric pressure, i.e., $T_{s_{11}} < 338$ K. Hence, assuming that the methanol composition at the top is close to one, it would be difficult for an equilibrium condenser or for the condenser proposed here, without subcooled product, to fit this temperature. Consequently, in the parameter estimation, the measurement coming from the condenser is weighted in a smaller proportion with respect to the rest of measurements. On the other hand, the parameter estimation is formulated such that the temperature of the liquid bulk phase of the model fits the measurement data. The vapor phase is not used here since measured profiles seems to adjust better to a subcooled liquid phase than to a vapor phase. The physical explanation is

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related to the fact that condensed liquid does not directly detaches from the

spiral condenser after condensation, but runs down the spiral before falling

down and wetting the temperature sensor.

5.3. Model calibration and validation

Figure 3 depicts the normalized sensitivity of the vapor and liquid tem-

peratures with respect to model parameters along with sensitivities for the

liquid composition at top and bottom stages. On the one hand, the heat loss

coefficients $(\psi_L, \psi_V, \psi_R \text{ and } \psi_C)$ and feed composition (x_F) exhibit a consid-

erable effect over the steady state profiles while the mass transfer coefficients

 $(C_{k^L} \text{ and } C_{k^V})$ have less influence in the steady state temperature profiles.

On the other hand, it is clear that volumes in the reboiler and condenser (\bar{v}_R)

and \bar{v}_C) along with pressure drop $(C_{\Delta p})$ and liquid holdup coefficients (C_h)

cannot be estimated from the steady state temperature measurements. As

this study only considers a pilot-scale column that is made out of glass and is

not insulated for educational reasons, sensitivity results can be different for

industrial columns. Nevertheless a sensitivity analysis will in general always

reveal the most informative parameters.

Hence, only parameters ϕ_L , ϕ_V , ϕ_R , ϕ_D , C_{kL} , C_{kV} and x_F have to be tuned.

To this end the deviations between the predicted temperatures for the liq-

uid bulk phase and the measured temperatures is minimised. Initial values,

bounds and obtained estimates are presented in Table 1. The experiments

are performed under different conditions, allowing for different feed com-

positions, x_F . Consequently, a different value of x_F is estimated for each

experiment. This increases the number of parameters again to $N_p = 6 + M$

where M is the number of experiments used for identification. Due to the structure of the setup, N is selected equal to 20. Five experiments are used for the estimation task (M = 5), leading to $N_p = 11$ and an optimization problem of size $1351 (M(14N - 12) + N_p)$.

Figure 4 (top) illustrates the fit results for the identification set, while Figure 4 (bottom) displays the results for the validation set. Although the estimation problem is nonlinear, a simultaneous approach estimating both the states and parameters at the same time allows to initialise also the states based on the available measurements and, hence, reduces the chances of getting stuck in a local minimum. Note that there is a group of points that lies outside de ± 3 K band around 335 K. These are measurements obtained from the condenser which cannot be totally explained by the model due to the inability to model subcooled liquid at this place. In the real setup, the liquid stream leaving the condenser is subcooled when falling along the spiral condenser. This liquids falls into the reflux drum decreasing its temperature. Since the model assumes that the liquid coming out from the condenser is at equilibrium, it cannot reach temperatures that the modified Rault's law does not predict. Hence, the only form of obtaining a subcooled liquid in the drum is increasing the heat loss coefficient in the reflux drum ϕ_C . Consequently, the heat that is removed by the subcooling of the liquid falling in the real condenser is compensated by the reflux drum losses in the model. A better representation of what is happening in the condenser can be achieved by using a non-equilibrium stage. This implies, however, increasing the complexity of the condenser model, since holdups for the non-equilibrium condenser have

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to be estimated from its geometry. This estimation is in general not trivial

and the current model already provides an acceptable prediction.

Figure 5 displays the boiling point diagrams for the validation set with

experiments 6 and 7 from Table 3. This figure illustrates the consistency of

the results, i.e., pairs composition-temperature for vapor (y^V, T^V) and liquid

 (x^L, T^L) bulk phases lie either in the superheated region or the subcooled

region, respectively.

6. Conclusions

In this paper a rigorous rate based model for separation in packed columns

has been calibrated based on multiple experiments. As a simultaneous ap-

proach has been used, an efficient computational procedure to deal with the

large number of degrees of freedom is required. Model reduction and struc-

ture exploitation have been employed, yielding a speedup by a factor 20. In

addition, the number of parameters has been reduced via a sensitivity anal-

ysis. Also additional constraints to ensure feasibility of the solution have

been included in the simultaneous approach in a natural way. As a result,

the calibrated model obtained with the presented procedure based on five

experiments, has been successfully validated on two additional experiments.

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References

References

Barroso, J., Borges, J., Oliveira, P., Pinheiro, C., Pires, A.C., Silva, J.M.

2009. Nonlinear modeling of a real pilot scale continuous distillation pro-

cess. Computer Aided Chemical Engineering. 28, 1733 - 1738.

Bonilla, J., 2011. Structure and convexity exploitation in nonlinear chemi-

cal process modeling and estimation. Ph.D. thesis, Chemical Engineering

Department. Katholieke Universiteit Leuven. Belgium.

Bonilla, J., Logist, F., Degrève, J., De Moor, B., Van Impe, J., 2012. A re-

duced order rate based model for distillation in packed columns: Dynamic

simulation and the differentiation index problem. Chemical Engineering

Science (68), 401 - 412.

Chambel, A.J.S., Pinheiro, C.I.C., Borges, J., Silva, J.M., 2011. Method-

ologies for input-output data exchange between LabVIEW and MAT-

LAB/Simulink software for Real Time Control of a Pilot Scale Distillation

Process. Computer Aided Chemical Engineering. 29, 708 - 712.

Dongarra, J. J., Duff, I. S., Sorensen, D. C., van der Vorst, H. A., 1988.

Numerical linear algebra for high-performance computers. SIAM.

Khrishnamurthy, R., Taylor, R., 1985. A nonequilibrium stage model for multicomponent separation processes: Part I model description and method of solution. AICHE Journal. 31 (3), 449–456.

Nocedal, J., Wright, S., 2006. Numerical optimization, 2nd Edition. Springer, New York.

Seader, J. D., Henley, E. J., 2006. Separation process principles. Wiley & Sons, Inc.

Taylor, R., Krishna, R., 1993. Multicomponent mass transfer. Wiley & Sons, Inc., New York.

Figures

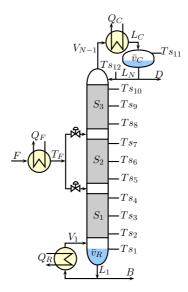


Figure 1:

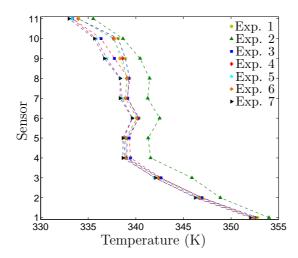


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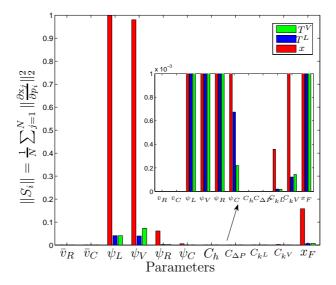


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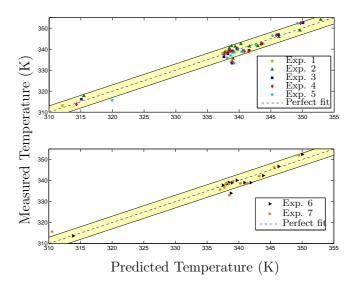


Figure 4:

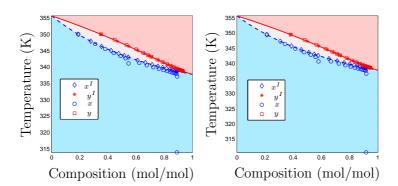


Figure 5:

Figure captions

Figure 1: General layout of the distillation setup.

Figure 2: Experimental data from the 11 temperature sensors along the column. Temperature of the reflux, around 315 K, has not been included in the plot but is used in the estimation.

Figure 3: Norm on the normalized sensitivity of the states with respect to the parameters.

Figure 4: Results for five calibration experiments (top) and two validation experiments (bottom).

Figure 5: Boiling point diagram for the mixture of Methanol-Isopropanol with states given by the steady state solution of the RBM for experiment 6 (left) and 7 (right).

Table 1: Parameter vector: initial (p_0) and optimised (p^*)

Parameter	Initial value	Bounds	Optimum ^a	Units	
\bar{v}_R	5	[3 6]	-	1	
$ar{v}_C$	2	[1 3]	-	1	
ϕ_L	1	$[0 \ 10]$	2.3	W/K	
ϕ_V	0.5	$[0 \ 10]$	0.0	W/K	
ϕ_R	8	[0 100]	0.0	W/K	
ϕ_C	5	[0 100]	8.3	W/K	
C_h	1	$[0 \ 2]$	-		
C_p	1	$[0 \ 2]$	-		
C_{k^L}	1	$[0 \ 5]$	0.22		
C_{k^V}	1	$[0 \ 5]$	2.12		
x_F	0.67	$[0.4 \ 0.7]$	0.4670	$\mathrm{mol/mol}$	

 $[^]a {\rm Only}$ identifiable parameters are optimized.

Table 2: Performance comparison between the full rate based model (FRBM) and the reduced rate based model (RORBM) when sparsity is exploited (SP) or not (NSP).

	FRBM-NSP	FRBM-SP	RORBM-NSP	RORBM-SP
Number of variables	383	383	272	272
Function evaluations	2688	161	1638	108
Execution time (s)	17.6	1.89	11.96	0.883
Memory used ^{a} (kB)	1146	4.36	578	2.73

 $[^]a$ Only the number of kilobytes to store the Jacobian in double precision format are considered as indication of the memory usage.

Table 3: Steady state experiments used for identification and validation.

Exp	Q_R	L_N	$F_{\rm in}$	T_F	$T_{ m amb}$	D	B
1	4.0	60.0	150	313.15	292.90	70	80
2	4.5	86.0	110	313.15	292.95	70	40
3	4.5	80.0	150	318.15	292.45	70	80
4	4.0	59.0	150	313.15	295.20	70	80
5	4.5	76.5	150	313.15	298.85	70	80
6	4.0	65.0	150	313.15	294.95	70	80
7	4.5	77.2	150	313.15	287.45	70	80